

# Gr. Helium Excited State and Origin of Magnetism: Brief Overview

- Picture: one electron in  $\phi_{1s}$ , another in  $\phi_{2s}$

could be

$2s$	$\uparrow$	$\downarrow$	$\downarrow$	$\uparrow$
$1s$	$\uparrow$	$\downarrow$	$\uparrow$	$\downarrow$

Spatial Part

Spin Part

one  
state  
( $S=0$ )

$$\frac{1}{\sqrt{2}} [\phi_{1s}(\vec{r}_1) \phi_{2s}(\vec{r}_2) + \phi_{1s}(\vec{r}_2) \phi_{2s}(\vec{r}_1)] \cdot \frac{1}{\sqrt{2}} [\alpha(1)\beta(2) - \alpha(2)\beta(1)] \quad (44)$$

(symmetric) (anti-symmetric) [singlet]

three  
states  
( $S=1$ )

$$\frac{1}{\sqrt{2}} [\phi_{1s}(\vec{r}_1) \phi_{2s}(\vec{r}_2) - \phi_{1s}(\vec{r}_2) \phi_{2s}(\vec{r}_1)] \cdot \begin{cases} \alpha(1)\alpha(2) \\ \frac{1}{\sqrt{2}} [\alpha(1)\beta(2) + \alpha(2)\beta(1)] \\ \beta(1)\beta(2) \end{cases} \quad (45)$$

(anti-symmetric) (symmetric) [triplet]

Aside:

Using Eqs. (44) and (45) to set up self-consistency equations to solve for  $\phi_{1s}(\epsilon_{1s})$  and  $\phi_{2s}(\epsilon_{2s})$ , one gets the Hartree-Fock approximation. The approximation takes proper account of the anti-symmetric requirement of many-electron wavefunctions.

Will  $S=0$  state and  $S=1$  states have different energies?

$$\hat{H}_{\text{He}} = \hat{h}_1 + \hat{h}_2 + \frac{e^2}{4\pi\epsilon_0 |\vec{r}_1 - \vec{r}_2|} \quad (10)$$

$S=0$  state

$$E_{\text{singlet}} = \iint \frac{1}{2} [\phi_{1s}^*(\vec{r}_1) \phi_{2s}^*(\vec{r}_2) + \phi_{1s}^*(\vec{r}_2) \phi_{2s}^*(\vec{r}_1)] \hat{H}_{\text{He}} [\phi_{1s}(\vec{r}_1) \phi_{2s}(\vec{r}_2) + \phi_{1s}(\vec{r}_2) \phi_{2s}(\vec{r}_1)] d^3r_1 d^3r_2 \quad (46)^+$$

$S=1$  state

$$E_{\text{triplet}} = \iint \frac{1}{2} [\phi_{1s}^*(\vec{r}_1) \phi_{2s}^*(\vec{r}_2) - \phi_{1s}^*(\vec{r}_2) \phi_{2s}^*(\vec{r}_1)] \hat{H}_{\text{He}} [\phi_{1s}(\vec{r}_1) \phi_{2s}(\vec{r}_2) - \phi_{1s}(\vec{r}_2) \phi_{2s}(\vec{r}_1)] d^3r_1 d^3r_2 \quad (47)^+$$

[They are not the same, by inspection]

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+ Where does the spin part go?

Roughly,

$$E_{\text{singlet}} = \underbrace{[\text{some terms}]}_{I \ \& \ J_{1s,2s}} + \text{terms like } \iint \phi_{1s}^*(\vec{r}_1) \phi_{2s}^*(\vec{r}_2) \frac{e^2}{4\pi\epsilon_0 |\vec{r}_1 - \vec{r}_2|} \phi_{1s}(\vec{r}_2) \phi_{2s}(\vec{r}_1) d^3r_1 d^3r_2$$

(48)

called  $K_{1s,2s}$

$$E_{\text{triplet}} = \underbrace{[\text{some terms}]}_{I \ \& \ J_{1s,2s}} - \text{terms like } \iint \phi_{1s}^*(\vec{r}_1) \phi_{2s}^*(\vec{r}_2) \frac{e^2}{4\pi\epsilon_0 |\vec{r}_1 - \vec{r}_2|} \phi_{1s}(\vec{r}_1) \phi_{2s}(\vec{r}_2) d^3r_1 d^3r_2$$

- $K_{1s,2s}$  has no classical analogy [inspect states and coordinates] (entirely quantum)
- called exchange Coulomb integral (or indirect Coulomb integral)

For helium atom,  $E_{\text{triplet}} < E_{\text{singlet}}$

$\Rightarrow$  S=1 is a lower energy excited state than S=0

$$K_{a,b} \equiv \int \phi_a^*(\vec{r}_1) \phi_b^*(\vec{r}_2) \frac{e^2}{4\pi\epsilon_0|\vec{r}_1 - \vec{r}_2|} \phi_a(\vec{r}_2) \phi_b(\vec{r}_1) d^3r_1 d^3r_2 \quad (49)$$

• entirely quantum effect      • Exchange Coulomb Integral<sup>†</sup>  
 compared with

$$J_{a,b} \equiv \int \phi_a^*(\vec{r}_1) \phi_b^*(\vec{r}_2) \frac{e^2}{4\pi\epsilon_0|\vec{r}_1 - \vec{r}_2|} \phi_a(\vec{r}_1) \phi_b(\vec{r}_2) d^3r_1 d^3r_2 \quad (50)$$

• has classical interpretation      • Direct Coulomb Integral

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<sup>†</sup>  $K_{a,b}$  arises because of the anti-symmetric requirement of many-electron wavefunctions

1<sup>st</sup> excited state is S=1 state

Physical Reason: S=1 has  $\psi_{\text{spatial}}^{(\text{anti-sym})} = \frac{1}{\sqrt{2}} \left[ \underbrace{\phi_{1s}(\vec{r}_1) \phi_{2s}(\vec{r}_2) - \phi_{1s}(\vec{r}_2) \phi_{2s}(\vec{r}_1)}_{\text{can't find electrons at the same place!}} \right]$   
 this is zero for  $\vec{r}_1 = \vec{r}_2$   
 antisymmetric

$\psi_{\text{spatial}}^{(\text{anti-sym})}(\vec{r}_1, \vec{r}_2)$  forbids electrons to be at the same place

Important idea!

Good!

lowers Coulomb Repulsion  $\frac{e^2}{4\pi\epsilon_0 |\vec{r}_1 - \vec{r}_2|}$  between them  
 because electrons can't get too close!

[In contrast,  $\psi_{\text{spatial}}^{(\text{sym})}$  (S=0 state) allows electrons to be close, higher energy!]  
 (likes)



▪ This is the physics behind the Hund's rule<sup>†</sup>  
 2 electrons  
 in p states:  $\uparrow \uparrow \_$  or  $\uparrow\downarrow \_ \_$

Hund's rule prefers this [also works for electrons in d and f states]

Why?

- $\Psi_{spin}$  is symmetric  $\Rightarrow \Psi_{spatial}$  is anti-symmetric
- $\Rightarrow$  avoid electrons to be at same place
- $\Rightarrow$  suppress Coulomb repulsion
- $\Rightarrow$  lower energy, thus preferred!

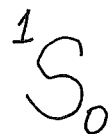
This is really "Thinking like a physicist"!

<sup>†</sup> Extending to many-electron cases is more involved. But conclusion remains. Or else, do QM calculation with Slater Determinant.

## Recall: Term Symbol

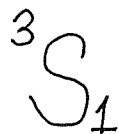
### Helium atom

- Ground State [two electrons in 1s, spin ( $S=0$ ) singlet]



- 1<sup>st</sup> excited state [one electron in 1s, another in 2s, spin ( $S=1$ ) triplet]

$$1s \Rightarrow l=0, \quad 2s \Rightarrow l=0 \quad \therefore \underbrace{L=0}$$



the "S" in the center

$$S=1, \quad L=0 \Rightarrow \underline{J=1}$$

the "1" in lower corner

$$(2S+1 = \underline{3})$$





## Let there be Magnets

- Atoms (some, not all) are like a small magnetic  
 [Fe, Co, Ni have aligned electrons' spins in 3d]
- It is spins, thus  $\vec{\mu}_s$ , that matter  
 [spin is very quantum]
- Ferromagnetic materials:  $\vec{\mu}$ 's of neighboring atoms aligned
  - Mechanism is NOT due to classical EM dipole-dipole interaction  
 [can't explain why Fe, Co, Ni, Gd are ferromagnetic at room temp.]
  - Mechanism is quantum (due to exchange Coulomb integral)  
 $\therefore$  due to electron-electron  $\frac{e^2}{4\pi\epsilon_0|\vec{r}_1-\vec{r}_2|}$  interaction

